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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/698,818	10/31/2003	Levi T. Thompson	2115-002542 2440 EXAMINER	
27572	7590 09/20/2004			
HARNESS, DICKEY & PIERCE, P.L.C. P.O. BOX 828			HAILEY, PATRICIA L	
BLOOMFIELD HILLS, MI 48303			ART UNIT	PAPER NUMBER
			1755	
			DATE MAILED: 09/20/2004	į

Please find below and/or attached an Office communication concerning this application or proceeding.

	Application No.	Applicant(s)		
Office Action Summary	10/698,818	THOMPSON ET AL.		
Onice Action Summary	Examiner	Art Unit		
The MAN INO DATE And	Patricia L. Hailey	1755		
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address		
A SHORTENED STATUTORY PERIOD FOR REPLY THE MAILING DATE OF THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication.  - If the period for reply specified above is less than thirty (30) days, a reply - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, however, may a reply be ting within the statutory minimum of thirty (30) day fill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	nely filed  rs will be considered timely.  the mailing date of this communication.  ID (35 U.S.C. § 133)		
Status				
1) Responsive to communication(s) filed on 20 Au	igust 2004.			
<u> </u>	action is non-final.			
3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is				
closed in accordance with the practice under E	x parte Quayle, 1935 C.D. 11, 45	53 O.G. 213.		
Disposition of Claims				
<ul> <li>4)  Claim(s) 1-29 is/are pending in the application.</li> <li>4a) Of the above claim(s) 14-21 is/are withdraw</li> <li>5)  Claim(s) is/are allowed.</li> <li>6)  Claim(s) 1-4,6-13 and 22-29 is/are rejected.</li> <li>7)  Claim(s) 5 is/are objected to.</li> <li>8)  Claim(s) are subject to restriction and/or</li> </ul>				
Application Papers				
9) The specification is objected to by the Examiner 10) The drawing(s) filed on is/are: a) access applicant may not request that any objection to the conference of the	epted or b) objected to by the liderawing(s) be held in abeyance. See on is required if the drawing(s) is obj	e 37 CFR 1.85(a). jected to. See 37 CFR 1.121(d).		
Priority under 35 U.S.C. § 119				
12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of:  1. Certified copies of the priority documents 2. Certified copies of the priority documents 3. Copies of the certified copies of the priori application from the International Bureau * See the attached detailed Office action for a list of	s have been received. s have been received in Application ity documents have been received (PCT Rule 17.2(a)).	on No ed in this National Stage		
Attachment(s)  1) Notice of References Cited (PTO-892)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  3) Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date 2/10/04.	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:			

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#### Election/Restrictions

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1. Applicant's election with traverse of Group I, claims 1-13 and 22-29, in the reply filed on August 20, 2004, is acknowledged. The traversal is on the ground(s) that maintaining both groups of claims in the same application "would not cause undue hardship for the Examiner". This is not found persuasive because the Examiner's reasons for the restriction requirement are not due to "undue hardship". In the restriction requirement, the claims were shown to be independent and distinct. An inventive water gas shift reaction is considered independent and distinct from a catalyst employed in said reaction, as well as a method of producing said catalyst. With respect to Applicants' claims, the elected "catalyst composition" and method for preparing a "supported transition metal composition" is independent from the non-elected water gas shift reaction, because Applicants have not shown how "the catalyst composition and method of preparing a supported transition metal composition claims generally parallel the method for carrying out a water gas shift reaction claims".

For these reasons, the requirement is still deemed proper and is therefore made FINAL.

2. Claims 14-21 are withdrawn from further consideration pursuant to 37 CFR 1.142(b), as being drawn to a nonelected method for carrying out a water gas shift reaction, there being no allowable generic or linking claim. Applicant timely

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traversed the restriction (election) requirement in the reply filed on August 20, 2004.

# Claim Rejections - 35 USC § 102

3. The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless -

- (b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.
- 4. Claims 1-4, 7-13, and 22-29 are rejected under 35 U.S.C. 102(b) as being anticipated by Thompson et al. (U. S. Patent No. 6,297,185).

Thompson et al. disclose a catalyst comprised of a support body that is a transition metal based, electrically conductive ceramic, and of at least one noble metal supported on the support body. The ceramic comprises a compound of at least one transition metal, the compound being selected from the group consisting of carbides, nitrides, borides, silicides, and combinations thereof. The noble metal may comprise a single metal, or may be an alloy of metals. Exemplary noble metals include platinum, palladium, osmium, iridium, gold, ruthenium, and rhodium. See col. 2, lines 24-39 and col. 1, lines 26-34 of Thompson et al.

The transition metals in the electrically conductive ceramic most preferably comprise transition metals from Groups III-VII; most specifically preferred

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transition metals include those from Groups V and VI. See col. 3, lines 6-19 of Thompson et al.

The Examples of Thompson et al. show a catalyst comprising approximately 10% platinum supported on tungsten carbide, as well as 6.2% platinum and 3.1% ruthenium on tungsten carbide (Examples 1 and 3), and further show that carbides, as well as nitrides, of transition metals such as molybdenum, are suitable ceramic supports (Example 4). The Examples further show exemplary methods of making the catalysts, wherein chloroplatinic acid is dissolved in water to produce a solution having a pH of about 2, and sodium bisulfite is added to raise the pH to about 3.5. (Optionally, a second metal can be incorporated, e.g., ruthenium; if such is the case, ruthenium trichloride is dissolved in water and added to the platinum solution). The pH is adjusted to approximately 5 via addition of sodium carbonate, and hydrogen peroxide is added to adjust the pH to between 3 and 5, and the solution is allowed to stand to permit excess peroxide to decompose. The ceramic support is then added to the solution to produce a slurry, which is subjected to stirring and hydrogen bubbling to reduce the metallic ions into free metal that precipitates onto the surface of the ceramic.

Another exemplary method involves dissolving hexachloroplatinic acid in methanol, followed by adding thereto a tungsten carbide ceramic powder. The methanol is removed by evaporation under vacuum to produce a dry powder. The powder is placed in a tube, through which a mixture of hydrogen and nitrogen is

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flowed (considered to read upon the limitation "absence of oxygen" in claim 29). The tube is heated to 110°C, and then to 400°C, at which reduction of the chloroplatinic acid takes place. Following reduction, a passivating atmosphere comprising 1% oxygen and the balance nitrogen and a temperature adjustment to 350°C is created within the tube.

In view of these teachings, Thompson et al. anticipate claims 1-4, 7-13, and 22-29.

# Claim Rejections - 35 USC § 103

- 5. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
  - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 6. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
  - 1. Determining the scope and contents of the prior art.
  - 2. Ascertaining the differences between the prior art and the claims at issue.
  - 3. Resolving the level of ordinary skill in the pertinent art.
  - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

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- 7. This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).
- 8. Claim 6 is rejected under 35 U.S.C. 103(a) as being unpatentable over Thompson et al. (U. S. Patent No. 6,297,185).

Thompson et al. is relied upon for its teachings stated in the above 102(b) rejection. While this reference teaches that gold is a known noble metal (col. 1, lines 32-34), the reference does not specifically teach the limitations of claim 6, namely a catalyst comprising gold on a support comprising a material of the formula recited in claim 1.

However, Thompson et al. at col. 4, lines 1-11 teach that Patentees' "present invention may be employed with any noble meal (sic) catalyst system."

(Emphasis added by Examiner). Thus, it would have been obvious to one of ordinary skill in the art to employ gold as a noble metal to be supported on an electrically conductive ceramic, such as carbides or nitrides of transition metals

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such as molybdenum or tungsten, and thereby obtain Applicants' claimed catalyst composition.

### Allowable Subject Matter

- 9. Claim 5 is objected to as being dependent upon a rejected base claim, but would be allowable if rewritten in independent form including all of the limitations of the base claim and any intervening claims.
- 10. The following is a statement of reasons for the indication of allowable subject matter:

Thompson et al. do not teach or suggest the limitations of claim 5, namely a catalyst comprising nickel on a support comprising a material of the formula recited in claim 1.

#### Conclusion

The prior art made of record and not relied upon is considered pertinent to 11. applicant's disclosure.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Patricia L. Hailey whose telephone number is (571) 272-1369. The examiner can normally be reached on Mondays-Thursdays.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark L. Bell can be reached on (571) 272-1362. The fax

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phone number for the organization where this application or proceeding is assigned

is 703-872-9306.

Any inquiry of a general nature or relating to the status of this application or

proceeding should be directed to the Group 1700 Receptionist, whose telephone

number is (571) 272-1700.

Information regarding the status of an application may be obtained from the

Patent Application Information Retrieval (PAIR) system. Status information for

published applications may be obtained from either Private PAIR or Public PAIR.

Status information for unpublished applications is available through Private PAIR

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Should you have questions on access to the Private PAIR system, contact the

Electronic Business Center (EBC) at 866-217-9197 (toll-free).

Patricia L. Hailey/plf

Examiner, Art Unit 1755

September 14, 2004

Mark I., Bell

Supervisory Patent Examiner

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